

## **APPLICATION FOR LETTERS PATENT**

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### **TREATMENT OF ACNE**

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## "Treatment of Acne"

## FIELD OF THE INVENTION

The invention relates to the use of antimicrobial metals for the treatment of acne.

## **BACKGROUND OF THE INVENTION**

Acne is a group of diseases whose initial pathology is the comedo and includes acne vulgaris, neonatal acne, infantile acne, and pomade acne. There are approximately 45 million people who suffer from acne in America alone. The disease is so common in youth at their puberty that it often has been termed physiological. Although acne stops appearing for most people by the age of 25, some people, the majority of them are women, experience the disease well into their adult life. This "adult acne" differs from teenage acne in location and that it tends to be more inflammatory with fewer comedones. It has received a great deal of attention recently.

Acne is a chronic inflammatory disease affecting the sebaceous glands. Acne lesions primarily involve the sebaceous glands located on the face, neck, chest and back. Both closed comedones (blackheads) and open comedones (whiteheads) are caused by hyperkeratinization of the infundibulum of the sebaceous duct. These keritinous plugs block the flow of sebum. These dilated ducts abound with the colonies of *Priopionibacterium acnes* and other fat splitting organisms. The clinically evident open and closed comedones and the microscopic microcomedo are the signal lesions of acne. The acne process results from a cascade of events. First, at puberty a spike in androgen production heralds an increase in sebum production and begins the hyperkeratinization process causing microcomedones and sebum blockade. With this blockage, the number of resident follicular flora increases dramatically. These bacteria produce inflammatory products, which permeate through thin walls of dilated sebum-filled duct. Once in the perifollicular dermis, they trigger the body's own immune defenses (both acute and granulomatous) to produce the characteristic inflammatory papules, pustules and nodules characteristic of inflammatory acne.

The term "acne" is used herein as a general term to include inflammatory diseases of the pilosebaceous unit. In the medical field, the specific type of acne is usually indicated by a modifying term, although the term acne is frequently used alone to designate common acne or acne vulgaris.

There are four factors that are generally believed to be the contributors to the development of acne, any of which evidence "an acne problem area":

- #### **Increased sebum production;**

- 1      2. Comedo formation, in which the follicular infundibulum hypercornifies,
- 2      hyperkeratinizes, and hypodesquamates;
- 3      3. Colonization of the follicle by anaerobic *Propionibacterium*, mainly *P. acnes*; and
- 4      4. The host's inflammatory response.

These four factors are interrelated to each other. Sebum is comedogenic and causes inflammation by itself. The *Propionibacterium* has high lipolytic activity and liberates free fatty acids from sebum lipids. The free fatty acid has been shown to cause marked inflammation. The microorganisms also produce other extracellular enzymes such as proteases and hyaluronidases, and chemotactic factors, which may be important in the inflammatory process. Apart from these four factors, serum hormones, especially dehydroepiandrosterone sulfate, have been found to correlate with acne.

Acne commonly afflicts adolescents and young adults; however, there is growing number of patients who develop acne in their late twenties or thirties. There are data that suggest a familial or genetic tendency for patients to develop severe cystic or conglobate acne. Additionally, acne has been linked to endocrine disorders, especially those characterized by elevated levels of circulating testosterone or testosterone congeners. Exogenous agents that could exacerbate acne include medications e.g. iodides, anti-seizure, certain antibiotics and corticosteroids.

A number of dermatological disorders resemble acne having papules and pustules, but not comedones. These conditions represent follicular inflammation caused by drugs (androgens, lithium), bacteria, mechanical friction and unknown causes. Rosacea is the most common of these disorders and is receiving the most marketing attention with direct to consumer advertising by drug companies. This activity is directed toward the patients using topical metronidazole that appears to have anti-inflammatory activity. Treatment usually lasts 3 months.

Most acne treatments are directed at preventing inflammatory lesions, particularly the larger nodulo-cystic lesions that tend to be destructive and lead to permanent scarring. In general, visible comedones are the only minor cosmetic nuisances and do not lead to inflammatory lesions. Most acne treatment is directed to four arenas:

1. Keratinous plugs in sebaceous ducts;
2. Large sebaceous glands producing excess sebum;
3. Increased numbers of resident follicular bacteria; and

1       4. Inflammatory response to chemical mediators passing through the follicular wall.  
2              Topical products used to remove comedones are known as comedolytics, the most  
3              effective being tretinoin, marketed as a prescription product (Retin A) and by several generic  
4              companies. Tretinoin or all-trans retinoic acid is the naturally occurring metabolite of  
5              Vitamin A. Tretinoin increases epidermal cell turnover, thus causing comedolysis and most  
6              importantly prevents the formation of new keratinous plugs. Applications of tretinoin is  
7              normally once a day at bedtime. Dryness, stinging and redness sometimes accompany the  
8              applications. Importantly, improvement is usually not seen for 6-8 weeks. Adapalene 0.1%  
9              (Differin) is a topical retinoid like tretinoin. Available by prescription only, the gel is usually  
10             applied once nightly. Side effects include frequent scaling, burning, redness and dryness.  
11             Improvement is delayed and is not evident for 4-8 weeks. Sodium sulfacetamide 10%/sulfur  
12             5% (Sulfacet-R) is also available by prescription only. It is a lotion with antibacterial and  
13             comedolytic action. As with tretinoin, improvement is seen in 4-8 weeks. Salicylic acid 2%  
14             is an over the counter product that exhibits mild comedolytics activity.

15             The only products that have anti-sebum activity are estrogens and 13 cis-retinoic acid  
16             (isotretinoin) and these must be used systemically to be effective. Isotretinoin (Accutane) is a  
17             metabolite of Vitamin A available by prescription only. Isotretinoin is used to treat only  
18             severe cystic or conglobate acne. Because of its teratogenic properties, birth defects can  
19             occur. Isotretinoin is a powerful drug and can elevate triglycerides, total cholesterol and  
20             decrease high-density lipoproteins (HDL). Other side effects include dry skin, dry eyes,  
21             itching, headaches, nosebleed, and photosensitivity. It is generally taken for 4-5 months to  
22             see improvement. Recently, one brand of oral contraceptive has been approved for the  
23             treatment of acne for patients who request birth control.

24             A number of topical and systemic agents are used to lower the number of bacteria that  
25             colonize the follicular duct. These include benzoyl peroxide (BP), BP 5%/erythromycin 3%  
26             (Benzamycin). BP has antibacterial activity and drying effects and is available over the  
27             counter or by prescription. BP is applied once or twice daily for 1-2 months. BP can produce  
28             erythema and peeling of skin. BP is often tried first for both non-inflammatory and mild  
29             inflammatory acne. Other topical antibiotics include clindamycin and erythromycin. These  
30             are used as solutions, lotions or gels by prescription only. Usually they are applied once or  
31             twice daily and results are seen in 1-2 months. Azelaic acid 20% (Azelex) also has mild  
32             antibacterial effects.

1 Systemic antibiotics include tetracycline and its analogs, which are used in low doses  
2 for years or until the end of the acne prone years. Most patients with mild inflammatory acne  
3 receive a combination of topical antibiotics and tretinoin or other retinoid. Bacterial  
4 resistance does occur so antibiotics may be changed or BP is substituted since resistance does  
5 not occur with BP. More severe acne requires systemic antibiotics and topical retinoid. The  
6 most severe must receive oral isotretinoin for 4-5 months.

7 There are no drugs that directly affect the inflammatory acne. The retinoids do have  
8 some anti-inflammatory properties, but these are poorly described. Topical steroid and even  
9 systemic steroids have been used to abort a severe flare of fulminant acne, but these are  
10 limited uses because of the side effects. Benzoyl peroxide gels are sometimes used as first  
11 aid on acne lesions. These function as a "drawing poultice", but data supporting this use is  
12 not available.

13 The treatment for acne centers around opening the pore, killing *P. acnes*, reducing  
14 sebum production and regulating inflammatory responses. Retinoids are the agents to reduce  
15 sebum production and open the pore. As a topical agent, Differin (adapalene) or Retin-A  
16 (tretinoin) is used for mild and moderate acne. Isotretinoin, an oral drug, is very effective but  
17 reserved for the severe and resistant acne because of its teratogenicity, hepatotoxicity,  
18 elevating triglyceride level and other side effects.

19 Topical and/or oral antimicrobial agents are commonly employed to treat acne.  
20 Topical benzoyl peroxide is used the most widely and is considered the best single topical  
21 agent. Tetracycline, Doxycycline and Minocycline are among the first line choices of oral  
22 antibiotics, in which Minocycline remains the premier drug for the treatment-resistant acne.  
23 Besides the side effects of the antimicrobial agents, development of resistant microorganisms  
24 has become an important issue nowadays. The number of patients harboring resistant *P.*  
25 *acnes* has been shown to be growing. While the patent literature reports that silver metal or  
26 silver salts such as silver nitrate, silver halides or silver sulphadiazine are amongst useful  
27 antibacterial agents for acne treatment, they have not, to the inventor's knowledge, been  
28 widely adopted for acne treatment.

29 Other approaches to acne include hormonal therapy with oral contraceptive pills as the  
30 major choice. It brings benefits to some patients over long term application. Intralesional  
31 corticosteroids and topical nicotinamide have also been used to control host's inflammatory  
32 response.

## SUMMARY OF THE INVENTION

Through research, the inventors have established that crystalline anti-microbial metals such as silver, preferably formed with atomic disorder, are effective antimicrobial agents against the bacteria associated with acne. The inventors have further established through clinical observations with acne patients, and in animal experiments, that nanocrystalline anti-microbial metals such as silver, formed with atomic disorder, reduce the inflammatory reaction. This research has resulted in a new therapeutic treatment for acne. When the anti-microbial metal is silver, this new acne treatment will have advantages of fewer side effects and less chance of development of resistant bacteria.

The inventors have discovered that anti-microbial metals selected from one or more of silver, gold, platinum and palladium, are effective in the treatment of acne. These anti-microbial metals are formed with atomic disorder, such that ions, clusters, atoms or molecules of the metals are released at a concentration sufficient to provide a localized antimicrobial and anti-inflammatory effect. Most preferably, the antimicrobial metals are in a nanocrystalline form, and include sufficient atomic disorder to provide an antimicrobial and anti-inflammatory effect on a sustainable basis.

The crystalline forms of these antimicrobial metals may be used in any of the following formats:

- i) coatings of the anti-microbial metals on medical grade substrates, for example, dressings, fibres, and materials composed of for example polyethylene, high density polyethylene, polyvinylchloride, latex, silicone, cotton, rayon, polyester, nylon, cellulose, acetate, carboxymethylcellulose, alginate, chitin, chitosan and hydrofibres;
  - ii) gels, formulated with nanocrystalline powders or solutions of the antimicrobial metals with such materials as carboxymethylcellulose, alginate, chitin, chitosan and hydrofibres, together with such ingredients as pectin and viscosity enhancers;
  - iii) creams, lotions, pastes and ointments formulated with nanocrystalline powders or solutions of the antimicrobial metals, for example as emulsions or with drying emollients;
  - iv) liquids, formulated as solutions by dissolving nanocrystalline coatings or powders of the antimicrobial metals, for example as topical solutions, aerosols or mists;
  - v) powders, preferably prepared as nanocrystalline powders of the antimicrobial metals, or as nanocrystalline coatings of the antimicrobial metals on biocompatible substrates

1       in powder form, preferably on bioabsorbable and/or hygroscopic substrates such as:  
2           Synthetic Bioabsorbable Polymers: for example polyesters/polyactones such as  
3           polymers of polyglycolic acid, glycolide, lactic acid, lactide, dioxanone, trimethylene  
4           carbonate etc., polyanhydrides, polyesteramides, polyorthooesters, polyphosphazenes,  
5           and copolymers of these and related polymers or monomers, or  
6           Naturally Derived Polymers:

7           Proteins: albumin, fibrin, collagen, elastin;

8           Polysaccharides: chitosan, alginates, hyaluronic acid; and

9           Biosynthetic Polyesters: 3-hydroxybutyrate polymers.

10          In the above formats, the nanocrystalline antimicrobial metals are formulated from  
11       nanocrystalline coatings or nanocrystalline powders of the nanocrystalline antimicrobial  
12       metals, or from solutions prepared by dissolving the nanocrystalline coatings or powders  
13       therein. The formulations include a therapeutically effective amount of the coatings or  
14       powders, and most preferably, the following amounts:

15       For coatings:                           150 - 3000 nm thick coatings

16       For gels, creams and lotions: 0.01 - 5% by weight of the nanocrystalline  
17                                                   antimicrobial metal powder

18       For liquids                              0.001 - 1% by weight of the antimicrobial metal

19          Nanocrystalline coatings of the antimicrobial metals are most preferably deposited  
20       onto one or more layers of medical dressing materials which can be laminated with uncoated  
21       layers of medical dressing materials. The coatings can be prepared by known techniques for  
22       preparing nanocrystalline coatings, but are most preferably prepared by physical vapour  
23       deposition under conditions which create atomic disorder. The nanocrystalline coatings may  
24       be prepared to create an interference colour so as to provide an indicator, as described in prior  
25       patent application WO 98/41095, published September 24, 1998, and naming inventors R. E.  
26       Burrell and R. J. Precht.

27          Nanocrystalline powders of the antimicrobial metals may be prepared as  
28       nanocrystalline coatings, preferably of the above thickness, on powdered substrates such as  
29       chitin, or may be prepared as nanocrystalline coatings on a substrate such as a silicon wafer,  
30       and then scraped off as a nanocrystalline powder. Alternatively, fine grained or  
31       nanocrystalline powders of the antimicrobial metals may be cold worked to impart atomic  
32       disorder, as disclosed in prior patent application WO 93/23092, published November 25,

1 1993, naming Burrell *et al.*, as inventors.

2 As used herein and in the claims, the terms and phrases set out below have the  
3 meanings which follow.

4 "Metal" or "metals" includes one or more metals whether in the form of substantially  
5 pure metals, alloys or compounds such as oxides, nitrides, borides, sulphides, halides or  
6 hydrides.

7 "Antimicrobial metals" are silver, gold, platinum and palladium, or mixtures of such  
8 metals with same or other metals, with silver metal being the most preferred.

9 "Biocompatible" means generating no significant undesirable host response for the  
10 intended utility. Most preferably, biocompatible materials are non-toxic for the intended  
11 utility. Thus, for human utility, biocompatible is most preferably non-toxic to humans or  
12 human tissues.

13 "Sustained release" or "sustainable basis" are used to define release of atoms,  
14 molecules, ions or clusters of a antimicrobial metal that continues over time measured in  
15 hours or days, and thus distinguishes release of such metal species from the bulk metal, which  
16 release such species at a rate and concentration which is too low to be therapeutically  
17 effective, and from highly soluble salts of antimicrobial metals such as silver nitrate, which  
18 releases silver ions virtually instantly, but not continuously, in contact with an alcohol or  
19 electrolyte.

20 "Atomic disorder" includes high concentrations of: point defects in a crystal lattice,  
21 vacancies, line defects such as dislocations, interstitial atoms, amorphous regions, grain and  
22 sub grain boundaries and the like relative to its normal ordered crystalline state. Atomic  
23 disorder leads to irregularities in surface topography and inhomogeneities in the structure on  
24 a nanometer scale.

25 "Normal ordered crystalline state" means the crystallinity normally found in bulk  
26 metal materials, alloys or compounds formed as cast, wrought or plated metal products. Such  
27 materials contain only low concentrations of such atomic defects as vacancies, grain  
28 boundaries and dislocations.

29 "Diffusion", when used to describe conditions which limit diffusion in processes to  
30 create and retain atomic disorder, i.e. which freeze-in atomic disorder, means diffusion of  
31 atoms (adatom diffusion) and/or molecules on the surface or in the matrix of the material  
32 being formed.

1        "Alcohol or water-based electrolyte" is meant to include any alcohol or water-based  
2        electrolyte that the anti-microbial materials of the present invention might contact in order to  
3        activate (i.e. cause the release of species of the anti-microbial metal) into same. The term is  
4        meant to include alcohols, water, gels, fluids, solvents, and tissues containing water,  
5        including body fluids (for example blood, urine or saliva), and body tissue (for example skin,  
6        muscle or bone).

7        "Bioabsorbable" as used herein in association includes substrates which are useful in  
8        medical devices, that is which are biocompatible, and which are capable of bioabsorption in  
9        period of time ranging from hours to years, depending on the particular application.

10       "Bioabsorption" means the disappearance of materials from their initial application  
11       site in the body (human or mammalian) with or without degradation of the dispersed polymer  
12       molecules.

13       "Colour change" is meant to include changes of intensity of light under  
14       monochromatic light as well as changes of hue from white light containing more than one  
15       wavelength.

16       An "interference colour" is produced when light impinges on two or more partly  
17       reflective surfaces separated by a distance which bears the right relationship to the  
18       wavelength of the light to be removed by destructive interference.

19       "Partly reflective" when used to describe the base or top layer materials, means that  
20       the material has a surface which reflects a portion of incident light, but which also transmits a  
21       portion of the incident light. Reflection occurs when a ray of incoming light encounters a  
22       boundary or interface characterized by a change in refractive index between two media. For  
23       the top layer of the anti-microbial materials of this invention, that interface is with air. For  
24       the base layer, the interface is with the top layer. The reflectance of the base and top layers is  
25       balanced so as to generate an interference colour.

26       "Partly light transmissive" when used to describe a thin film of the top layer material  
27       means that the thin film is capable of transmitting at least a portion of incident visible light  
28       through the thin film.

29       "Detectable" when used to describe a colour change means an observable shift in the  
30       dominant wavelength of the reflected light, whether the change is detected by instrument,  
31       such as a spectrophotometer, or by the human eye. The dominant wavelength is the  
32       wavelength responsible for the colour being observed.

1        "Cold working" as used herein indicates that the material has been mechanically  
2        worked such as by milling, grinding, hammering, mortar and pestle or compressing, at  
3        temperatures lower than the recrystallization temperature of the material. This ensures that  
4        atomic disorder imparted through working is retained in the material.

5        "Therapeutically effective amount" is used herein to denote any amount of a  
6        formulation of the nanocrystalline antimicrobial metals which will exhibit an anti-microbial  
7        and an anti-inflammatory effect in acne when applied to the affected area. A single  
8        application of the formulations of the present invention may be sufficient, or the formulations  
9        may be applied repeatedly over a period of time, such as several times a day for a period of  
10      days or weeks. The amount of the active ingredient, that is the antimicrobial metal in the  
11      form of a coating, powder or dissolved in liquid solution, will vary with the conditions being  
12      treated, the stage of advancement of the condition, and the type and concentration of the  
13      formulation being applied. Appropriate amounts in any given instance will be readily  
14      apparent to those skilled in the art or capable of determination by routine experimentation.

15      "Nanocrystalline" is used herein to denote single-phase or multi-phase polycrystals,  
16      the grain size of which is less than about 100, more preferably < 50 and most preferably < 25  
17      nanometers in at least one dimension. The term, as applied to the crystallite or grain size in  
18      the crystal lattice of coatings, powders or flakes of the antimicrobial metals, is not meant to  
19      restrict the particle size of the materials when used in a powder form.

20      "Powder" is used herein to include particulate sizes of the nanocrystalline  
21      antimicrobial metals ranging from nanocrystalline powders to flakes. Most preferably,  
22      powders of the present invention are sized at less than 100  $\mu\text{m}$ , and more preferably less than  
23      40  $\mu\text{m}$ .

24      "Grain size", or "crystallite size" means the size of the largest dimension of the  
25      crystals in the antimicrobial metal coating or powder.

## 26      DESCRIPTION OF THE PREFERRED EMBODIMENTS

27      Crystalline forms of the antimicrobial metals Ag, Au, Pt, and Pd can be prepared as  
28      powders or coatings, or as solutions prepared by dissolving the coatings or powders. The  
29      crystalline coatings or powders are most preferably formed with atomic disorder in  
30      accordance with the techniques published in the prior patent applications of Burrell *et al.*, see  
31      for example WO 93/23092, published November 25, 1993, WO 95/13704, published May 26,  
32      1995 and WO 98/41095, published September 24, 1998.

1      A. Antimicrobial Metal Coatings on Dressings

2      Dressings carrying nanocrystalline coatings of antimicrobial metals in accordance  
3      with the invention include at least one, and possibly more layers of medical dressing  
4      materials. Multiple layers may be laminated together by known means such as low  
5      temperature thermal fusing, stitching or, most preferably, ultrasonic welding.

6      The dressing may be formed to include an occlusive or semi-occlusive layer such as  
7      an adhesive tape or polyurethane film in order to secure the dressing in place, and retain  
8      moisture for release of ions, atoms, molecules or clusters of the antimicrobial metal  
9      (hereinafter antimicrobial metal species).

10     The preferred and alternate compositions of the dressing layers, together with the  
11    preferred nanocrystalline antimicrobial metal coatings, are set out in further detail below.

12     I) Dressing Materials

13     The dressing is formed of a perforated, preferably non-adherent material which allows  
14    for fluids to penetrate or diffuse there through in either or both directions. The perforated  
15    material may be formed of a woven or non-woven, non-woven being preferred, fabric such as  
16    cotton, gauze, a polymeric net or mesh such as polyethylene, nylon, polypropylene or  
17    polyester, an elastomer such as polyurethane or polybutadiene elastomers, or a foam such as  
18    open cell polyurethane foam. Exemplary perforated, non-adherent materials useful for the  
19    dressing include non-woven meshes such as DELNET<sup>TM</sup> P530, which is a non-woven veil  
20    formed of high density polyethylene using extrusion, embossing and orientation processes,  
21    produced by Applied Extrusion Technologies, Inc. Of Middletown, Delaware, USA. This  
22    same product is available as Exu-Dry CONFORMANT 2<sup>TM</sup> Wound Veil, from Frass Survival  
23    Systems, Inc., Bronx, New York, USA as a subset of that company's Wound Dressing Roll  
24    (Non-Adherent) products. Other useful non-woven meshes include CARELLE<sup>TM</sup> or NYLON  
25    90<sup>TM</sup>, available from Carolina Formed Fabrics Corp., N-TERFACE<sup>TM</sup>, available from  
26    Winfield Laboratories, Inc., of Richardson, Texas, USA. Exemplary woven meshes may be  
27    formed from fibreglass or acetate, or cotton gauze. An exemplary hydrophilic polyurethane  
28    foam is HYPOL<sup>TM</sup>, available from W.R. Grace & Co., New York, NY, USA.

29     For ease of ultrasonic welding for lamination, at least one dressing layer is preferably  
30    formed from a polymeric material which is amenable to ultrasonic welding, that is which will  
31    melt on the application of localized heat and then fuse multiple layers together on cooling.

32     If desired, a second, absorbent layer is formed from an absorbent material for holding

1 sufficient moisture next to the skin in order to activate the antimicrobial metal coating, that is  
2 to cause release of ions, molecules, atoms or clusters of the antimicrobial metal in order to  
3 cause an anti-microbial and anti-inflammatory effect. Preferably, the absorbent material is an  
4 absorbent needle punched non-woven rayon/polyester core such as SONTARA™ 8411, a  
5 70/30 rayon/polyester blend commercially available from Dupont Canada, Mississauga,  
6 Ontario, Canada. This product is sold by National Patent Medical as an American White  
7 Cross sterile gauze pad. However, other suitable absorbent materials include woven or non-  
8 woven materials, non-woven being preferred made from fibers such as rayon, polyester,  
9 rayon/polyester, polyester/cotton, cotton and cellulosic fibers. Exemplary are creped  
10 cellulose wadding, an air felt of air laid pulp fibers, cotton, gauze, and other well known  
11 absorbent materials suitable for medical dressings.

12 A third layer of the dressing, if used, is preferably formed of perforated, non-adherent  
13 material such as used in the first layer. This allows moisture penetration as sterile water and  
14 the like are added in order to activate the antimicrobial metal coating.

15 Additional layers may be included between or above the first, second and third layers  
16 as is well known in medical dressings. The coated dressing layers may be combined with an  
17 adhesive layer, in a well known manner.

18 The dressing may be used as a single layer, or may be used as multiple layers  
19 laminated together at intermittent spaced locations across the dressing by ultrasonic welds.  
20 Ultrasonic welding is a known technique in the quilting art. Briefly, heat (generated  
21 ultrasonically) and pressure are applied to either side of the dressing at localized spots  
22 through an ultrasonic horn so as to cause flowing of at least one of the plastic materials in the  
23 first and second layers and the subsequent bonding together of the layers on cooling. The  
24 welds appear at localized circular spots and are preferably less than 0.5 cm in diameter.

25 The use of ultrasonic welding of the layers at spaced locations has the advantage of  
26 retaining the absorbent and moisture penetration properties of the dressing layers, while  
27 retaining the conforming properties of the dressing. Edge seams, stitching and adhesives  
28 have the disadvantage of interfering with one or more of these desirable properties of the  
29 dressings. Furthermore, by spacing the welds at intermittent locations across the dressing, the  
30 dressing may be cut to smaller sizes, as needed, without causing delamination. Preferred  
31 spacings of about 2.5 cm between welds allows the dressing to be cut down to about 2.5 cm  
32 sizes, while maintaining at least one weld to hold the laminated layers together.

1       ii) Nanocrystalline Coatings of Antimicrobial Metals

2              The dressing preferably includes a nanocrystalline coating of one or more of the  
3              antimicrobial metals. The coating is applied to one or more of the dressing layers, but is most  
4              preferably applied at least to the skin facing layer.

5              The nanocrystalline coating is most preferably formed with atomic disorder in  
6              accordance with the procedures set out above and as described in WO 93/23092, WO  
7              95/13704, and WO98/41095, and as set out below. Most preferably, the coating is formed as  
8              a multilayer coating of the antimicrobial metals, having a top and a base layer, as set below,  
9              to produce an interference colour. In this way, the coating provides not only the active  
10             ingredient for the treatment of acne, but also acts as an indicator of activation of the dressing.  
11             As the top layer of the coating is activated with an alcohol or water-based electrolyte, such as  
12             sterile water or ethanol, even minor dissolution of the antimicrobial metal results in a  
13             detectable colour change, indicating that the coating has been activated. If there is no colour  
14             change, additional moisture might be provided to the dressing by adding water, until a colour  
15             change is detected. Once activated, the dressing should be maintained in a moist condition,  
16             for example by the addition of sterile water, if necessary.

17       iii) Sterilization

18             Dressings with nanocrystalline coatings of a antimicrobial metal formed with atomic  
19             disorder are preferably sterilized without applying excessive thermal energy, which can  
20             anneal out the atomic disorder, thereby reducing or eliminating a useful release of  
21             antimicrobial metal species. Gamma radiation is preferred for sterilizing such dressings, as  
22             discussed in WO 95/13704. Electron beam and ethylene oxide sterilization techniques can  
23             also be used.

24             It should be appreciated that the use of ultrasonic welding to laminate the layers of  
25             dressings with nanocrystalline coatings formed from antimicrobial metals with atomic  
26             disorder is advantageous since it achieves bonding in localized spots and avoids applying heat  
27             to any significant portion of the dressing, thereby avoiding any significant reduction in the  
28             solubility of the antimicrobial metals through annealing out of the atomic disorder.

29             The sterilized dressings should be sealed in packaging which excludes light  
30             penetration to avoid additional oxidation of the antimicrobial metal coating. Polyester  
31             peelable pouches are preferred. The shelf life of dressings thus sealed is over one year.

1      iv) Directions for Use of Dressings for Acne

2            The dressing is placed on the affected portion of the skin and is moistened with drops  
3            of sterile water or, for example 70% ethanol, in order to activate the coating for release of  
4            antimicrobial metal species. The dressing is then secured in place with an occlusive or semi-  
5            occlusive layer, such as an adhesive tape or polyurethane film, which keeps the dressing in a  
6            moist environment.

7            As set out in Examples 5, dressings carrying a bi-layer nanocrystalline antimicrobial  
8            metal coating formed with silver having atomic disorder, manufactured as set out above and  
9            as described in greater detail in Example 1, have shown substantial clinical response in  
10          treating acne. Dressings prepared with a single layer of silver having a reduced content of  
11          oxygen in the coating, as set forth in Example 2, gave similar clinical results, without staining  
12          the skin. In use, the dressings are kept moist, at 100% relative humidity. Adding sterile  
13          water initially to activate the antimicrobial metal coating is needed, and then as needed to  
14          maintain the dressing in a moist condition. Dressings may be changed as required for  
15          observation and cleaning. Preferably dressings are changed daily, but could be left longer,  
16          such as 3 days, and can provide a therapeutic effect for a much longer period of time.

17      v) Multilayer Nanocrystalline Coatings of Antimicrobial Metals With Interference Colour

18            The dressings preferably include the antimicrobial metal coating formed with at least  
19          two metal layers, a base layer and a top layer over the base layer, so as to produce an  
20          interference colour, as set forth in WO 98/41095. Both layers are partly reflective; the top  
21          layer is partly light transmissive. The top layer is a thin film containing at least one  
22          antimicrobial metal formed with sufficient atomic disorder such that the top layer, in contact  
23          with an alcohol or water based electrolyte, releases ions, atoms, molecules or clusters of the  
24          antimicrobial metal, at a concentration sufficient to provide a therapeutic effect, on a  
25          sustainable basis. In this way, the top layer, in contact with the alcohol or electrolyte, will  
26          undergo a change in optical path length, either by a change in thickness resulting from some  
27          dissolution, or through a change in the refractive index of the top layer resulting from a  
28          change in the composition of a newly formed thin layer formed on the top layer. Either or  
29          both of these results are sufficient to cause a detectable colour change, thus providing an  
30          indicator that the top layer has been activated.

31            Both the base layer and the top layer are formed from a partly reflective material. In  
32          this way, at least a portion of the incoming light is reflected from the surface of the layer

1 while another portion is transmitted through the layer. The top layer is partly light  
2 transmissive to allow incident light to reach the interface with the base layer. The top layer  
3 thus cannot approximate 100% reflectivity, such as in pure Al or Ag. or interference colours  
4 cannot be generated, as is well known in the art. The materials for the top and base layers  
5 should be balanced in their reflectances in order to generate an interference colour.  
6 Generally, the top layer is deposited as a thin film having a thickness which maintains  
7 adequate transmittance to generate an interference colour. Furthermore, the refractive index  
8 for the materials in layers is different, accomplished by differences in their actual or effective  
9 compositions. For instance different materials in the two layers will result in the materials  
10 having different actual refractive indexes. However, if it is desired to make the layers from  
11 the same material, the layers can be deposited with different porosities or different  
12 levels/types of atomic disorder, in order to achieve different effective compositions, and thus  
13 different refractive indexes.

14 In this manner, incoming light reflects off the interface of the base and top layers.  
15 Incoming light reflects from the interface of the top layer with air, and interferes with the  
16 light reflected from the interface with the base layer so as to generate an "interference  
17 colour". The particular colour which is generated and its brightness will depend on the  
18 properties of the layers, most importantly on the composition of the layers, which determines  
19 its transmittance and absorption properties, along with its refractive index, and on the  
20 thickness of the layers. Generally, it is desirable to generate first and second order  
21 interference colours, by limiting the thickness of the base layer and top layers to minimize the  
22 number of internal reflections. First and second order interference colours are generally  
23 brighter than third and fourth order etc. colours, making them more aesthetically pleasing,  
24 more consistently reproducible in manufacturing, and more susceptible to detectable colour  
25 change on variations in thickness or dissolution of even a minor amount of the top layer.

26 The property which determines the particular colour which is generated is the  
27 effective optical thickness of the top layer, that is the product of the refractive index of the  
28 top layer material and the actual thickness of the top layer. Thus the colour which is desired  
29 can be altered by changing the actual thickness or the top layer or its refractive index.

30 Preferably, the material in the base layer is a reflective metal. Such metals are known  
31 in the art and include, for example one or more of the valve metals; e.g. Ta, Nb, Ti, Zr and  
32 Hf, as well as transition metals such as Au, Ag, Pt, Pd, Sn, Cu, V, W and Mo, or the metal Al.

1 More preferably, the base layer is formed from one or more of the antimicrobial metals Ag,  
2 Au, Pt, and Pd, most preferably Ag, in a partly reflective form.

3 The base layer may be formed by known techniques, such as the vapour deposition  
4 techniques of evaporation or physical vapour deposition. Preferably, the base layer is formed  
5 as a thin film by physical vapour deposition with atomic disorder, as set out below and in WO  
6 95/13704, in order to produce a sustainable release of the antimicrobial metal species when  
7 the base layer is ultimately exposed to an alcohol or water based electrolyte. The thickness  
8 of the base layer is generally not critical, provided that it is partly reflective. Preferred  
9 thicknesses will vary widely with the material composition. However, in that the layer is  
10 preferably a thin film formed by physical vapour deposition techniques, it should be at least  
11 about 25 nm thick to create a useful colour. The base layer should be greater than 60 nm  
12 thick, more preferably 300 to 2500 nm thick, and most preferably 600 to 900 nm thick.

13 The top layer is formed of a partly reflective, partly light transmissive thin film  
14 containing at least one antimicrobial metal, most preferably Ag, formed with atomic disorder  
15 so as to produce a sustainable release of the antimicrobial metal species, and ultimate colour  
16 change, when exposed to an alcohol or a water based electrolyte. The thickness of the top  
17 layer formed from these metals is preferably less than 400 nm in order to maintain the  
18 preferred level of light transmission. The desired thickness will vary with the composition of  
19 the top layer, and with the desired end colour and colour change. For first and second order  
20 interference colours, the thickness will generally be less than about 400 nm. More preferably,  
21 the thickness will range from 5 to 210 nm, most preferably from 10 to 100 nm.

22 The top layer may be a thin film of the base layer material, formed with a different  
23 refractive index for instance by altering the deposition conditions to change the porosity,  
24 composition and/or degree of atomic disorder in the layers.

25 When the base layer is itself formed from a antimicrobial metal with atomic disorder,  
26 the top layer may be provided as an in situ generated top layer by virtue of its thickness  
27 and/or composition changing on contacting an alcohol or water based electrolyte, so as to  
28 produce an interference colour which differs from the initial colour of the base layer.

29 Most preferably, the top layer is a thin film of a composite material formed by co-,  
30 sequentially or reactively depositing a antimicrobial metal in a matrix with atoms or  
31 molecules of a different material to create atomic disorder in the matrix, in the manner set out  
32 below. The different material is selected from a) biocompatible metals, b) oxygen, nitrogen,

1 hydrogen, boron, sulphur or halogens, or c) an oxide, nitride, carbide, boride, halide, sulphide  
2 or hydride of either or both of a antimicrobial metal or a biocompatible metal. Most  
3 preferably, the top layer material is a composite material containing silver, and one or both of  
4 silver oxide and atoms or molecules containing oxygen trapped or absorbed in the silver  
5 matrix. The term "silver oxide" is meant to include any oxide or mixture of oxides of silver.  
6 However, the top layer is preferably not formed solely of AgO and/or Ag<sub>2</sub>O, since the  
7 solubility of these materials is low.

8 vi) Nanocrystalline Coatings of Antimicrobial Metals Containing Atomic Disorder

9 At least the top layer, and preferably also the base layer, is formed in a crystalline  
10 form from one or more antimicrobial metals with atomic disorder. The production of atomic  
11 disorder through physical vapour deposition techniques is described in WO 93/23092 and  
12 WO 95/13704, and as outlined below.

13 The antimicrobial metal is deposited as a thin metallic film on one or more surfaces of  
14 the dressing by vapour deposition techniques. Physical vapour techniques, which are well  
15 known in the art, all deposit the metal from the vapour, generally atom by atom, onto a  
16 substrate surface. The techniques include vacuum or arc evaporation, sputtering, magnetron  
17 sputtering and ion plating. The deposition is conducted in a manner to create atomic disorder  
18 in the coating as defined above. Various conditions responsible for producing atomic  
19 disorder are useful. These conditions are generally those which one has been taught to avoid  
20 in thin film deposition techniques, since the object of most thin film depositions is to create a  
21 defect free, smooth and dense film (see for example J.A. Thornton, "Influence of Apparatus  
22 Geometry and Deposition Conditions on the Structure and Topography of Thick Sputtered  
23 Coatings," J. Vac. Sci. Technol., 11(4), 666-670, 1974).

24 The preferred conditions which are used to create atomic disorder during the  
25 deposition process include:

26 - a low substrate temperature, that is maintaining the surface to be coated at a  
27 temperature such that the ratio of the substrate temperature to the melting point of the metal  
28 (in degrees Kelvin) is less than about 0.5, more preferably less than about 0.35 and most  
29 preferably less than about 0.3; and optionally one or both of:

30 - a higher than normal working (or ambient) gas pressure, i.e. for vacuum  
31 evaporation: e-beam or arc evaporation, greater than 0.001 Pa (0.01 mT), gas scattering  
32 evaporation (pressure plating) or reactive arc evaporation, greater than 2.67 Pa (20 mT); for

1 sputtering: greater than 10 Pa (75 mT); for magnetron sputtering: greater than about 1.33 Pa  
2 (10 mT); and for ion plating: greater than about 26.67 Pa (200 mT); and  
3 - maintaining the angle of incidence of the coating flux on the surface to be  
4 coated at less than about 75°, and preferably less than about 30°.

5 For economic reasons, the thin metal film has a thickness no greater than that needed  
6 to provide release of antimicrobial metal species on a sustainable basis over a suitable period  
7 of time, and to generate the desired interference colour. Within the preferred ranges of  
8 thicknesses set out above, the thickness will vary with the particular metal in the coating  
9 (which varies the solubility and abrasion resistance), and with the degree of atomic disorder  
10 in (and thus the solubility of) the coating. The thickness will be thin enough that the coating  
11 does not interfere with the dimensional tolerances or flexibility of the device for its intended  
12 utility.

13 The therapeutic effect of the material so produced is achieved when the coating is  
14 brought into contact with an alcohol or a water based electrolyte, thus releasing metal ions,  
15 atoms, molecules or clusters. The concentration of the metal species which is needed to  
16 produce a therapeutic effect will vary from metal to metal.

17 The ability to achieve release of metal atoms, ions, molecules or clusters on a  
18 sustainable basis from a coating is dictated by a number of factors, including coating  
19 characteristics such as composition, structure, solubility and thickness, and the nature of the  
20 environment in which the device is used. As the level of atomic disorder is increased, the  
21 amount of metal species released per unit time increases. For instance, a silver metal film  
22 deposited by magnetron sputtering at  $T/T_m < 0.5$  and a working gas pressure of about 0.93 Pa  
23 (7 mT) releases approximately 1/3 of the silver ions that a film deposited under similar  
24 conditions, but at 4 Pa (30 mT), will release over 10 days. Films that are created with an  
25 intermediate structure (ex. lower pressure, lower angle of incidence etc.) have Ag release  
26 values intermediate to these values as determined by bioassays. This then provides a method  
27 for producing controlled release metallic coatings. Slow release coatings are prepared such  
28 that the degree of disorder is low while fast release coatings are prepared such that the degree  
29 of disorder is high.

30 For continuous, uniform coatings, the time required for total dissolution will be a  
31 function of film thickness and the nature of the environment to which they are exposed. The  
32 relationship in respect of thickness is approximately linear, i.e. a two fold increase in film

thickness will result in about a two fold increase in longevity.

It is also possible to control the metal release from a coating by forming a thin film coating with a modulated structure. For instance, a coating deposited by magnetron sputtering such that the working gas pressure was low (ex. 2 Pa or 15 mT) for 50% of the deposition time and high (ex. 4 Pa or 30 mTorr) for the remaining time, has a rapid initial release of metal ions, followed by a longer period of slow release. This type of coating is extremely effective on devices such as urinary catheters for which an initial rapid release is required to achieve immediate anti-microbial concentrations followed by a lower release rate to sustain the concentration of metal ions over a period of weeks.

The substrate temperature used during vapour deposition should not be so low that annealing or recrystallization of the coating takes place as the coating warms to ambient temperatures or the temperatures at which it is to be used (ex. body temperature). This allowable  $\Delta T$ , that the temperature differential between the substrate temperature during deposition and the ultimate temperature of use, will vary from metal to metal. For the most preferred metal, Ag, preferred substrate temperatures of -20 to 200°C, more preferably -10°C to 100°C are used.

Atomic order may also be achieved, in either or both of the base and top layers by preparing composite metal materials, that is materials which contain one or more antimicrobial metals in a metal matrix which includes atoms or molecules different from the antimicrobial metals.

The preferred technique for preparing a composite material is to co- or sequentially deposit the antimicrobial metal(s) with one or more other inert, biocompatible metals selected from Ta, Ti, Nb, Zn, V, Hf, Mo, Si, Al and alloys of these metals or other metal elements, typically other transition metals. Such inert metals have a different atomic radii from that of the antimicrobial metals, which results in atomic disorder during deposition. Alloys of this kind can also serve to reduce atomic diffusion and thus stabilize the disordered structure. Thin film deposition equipment with multiple targets for the placement of each of the antimicrobial and biocompatible metals is preferably utilized. When layers are sequentially deposited the layer(s) of the biocompatible metal(s) should be discontinuous, for example as islands within the antimicrobial metal matrix. The final weight ratio of the antimicrobial metal(s) to biocompatible metal(s) should be greater than about 0.2. The most preferable biocompatible metals are Ti, Ta, Zn and Nb. It is also possible to form the anti-microbial

1 coating from oxides, carbides, nitrides, sulphides, borides, halides or hydrides of one or more  
2 of the antimicrobial metals and/or one or more of the biocompatible metals to achieve the  
3 desired atomic disorder.

4 Another composite material may be formed by reactively co- or sequentially  
5 depositing, by physical vapour techniques, a reacted material into the thin film of the  
6 antimicrobial metal(s). The reacted material is an oxide, nitride, carbide, boride, sulphide,  
7 hydride or halide of the antimicrobial and/or biocompatible metal, formed in situ by injecting  
8 the appropriate reactants, or gases containing same, (ex. air, oxygen, water, nitrogen,  
9 hydrogen, boron, sulphur, halogens) into the deposition chamber. Atoms or molecules of  
10 these gases may also become absorbed or trapped in the metal film to create atomic disorder.  
11 The reactant may be continuously supplied during deposition for codeposition or it may be  
12 pulsed to provide for sequential deposition. The final weight ratio of reaction product to  
13 antimicrobial metal(s) should be greater than about 0.05. Air, oxygen, nitrogen and hydrogen  
14 are particularly preferred reactants, with oxygen being most preferred.

15 The above deposition techniques to prepare composite coatings may be used with or  
16 without the conditions of lower substrate temperatures, high working gas pressures and low  
17 angles of incidence previously discussed. One or more of these conditions are preferred to  
18 retain and enhance the amount of atomic disorder created in the coating.

19 **B. Powders of Antimicrobial Metals**

20 Powders of one or more antimicrobial metals formed with atomic disorder, most  
21 preferably in a nanocrystalline form, may be prepared either as pure metals, metal alloys or  
22 compounds such as metal oxides or metal salts, by vapour deposition, mechanical working, or  
23 compressing to impart atomic disorder, as set out below, or by the procedures set out in WO  
24 93/23092, WO 95/13704, or as otherwise known in the art. Mechanically imparted disorder  
25 is conducted by milling, grinding, hammering, mortar and pestle or compressing, under  
26 conditions of low temperature (i.e., temperatures less than the temperature of recrystallization  
27 of the material) to ensure that annealing or recrystallization does not take place.

28 Nanocrystalline powders may be prepared by preparing nanocrystalline coatings in the  
29 manner set out above, onto a substrate such as a cold finger or a silicon wafer, and then  
30 scraping off the coating to form a powder. Alternatively, nanocrystalline coatings such as set  
31 out above may be formed onto powdered substrates which are biocompatible. Particularly  
32 preferred substrates are bioabsorbable and/or hygroscopic powders such as chitin. Exemplary

1 bioabsorbable and/or hygroscopic powders are composed of :

2 Synthetic Bioabsorbable Polymers: for example polyesters/polyactones such as polymers of  
3 polyglycolic acid, glycolide, lactic acid, lactide, dioxanone, trimethylene carbonate etc.,  
4 polyanhydrides, polyesteramides, polyorthoesters, polyphosphazenes, and copolymers of  
5 these and related polymers or monomers.

6 Naturally Derived Polymers:

7 Proteins: albumin, fibrin, collagen, elastin;

8 Polysaccharides: chitosan, alginates, hyaluronic acid; and

9 Biosynthetic Polyesters: 3-hydroxybutyrate polymers.

10 Most preferably, powders of the present invention are sized at less than 100  $\mu\text{m}$ , and  
11 more preferably less than 40  $\mu\text{m}$ .

12 The prepared nanocrystalline powders may then be incorporated into or onto medical  
13 dressings or pharmaceutical formulations, by any methods known in the art. For example, the  
14 powders may be layered onto the substrates (dressings or powders), mechanically mixed  
15 within the fibres of the dressings, impregnated into dressings by physical blowing, or added  
16 to topical pharmaceutical composition ingredients.

17 The antimicrobial and anti-inflammatory effects of the nanocrystalline powder is  
18 achieved when the powder is brought into contact with an alcohol or a water-based  
19 electrolyte, thus releasing the antimicrobial metal ions, atoms, molecules or clusters.

20 Nanocrystalline powders may be sterilized as described above.

21 **C. Formulations and Dosages**

22 Typically, the nanocrystalline antimicrobial metals will be formulated from the active  
23 ingredient, namely nanocrystalline powders or coatings of the antimicrobial metals, or  
24 dissolved species from such powders or coatings, in the form of:

- 25 • coatings on medical dressings or biocompatible powdered substrates,
- 26 • powders included in medical dressings,
- 27 • topical pharmaceutical compositions such as gels, pastes, ointments, creams, lotions,  
emulsions, suspensions or powders,
- 28 • liquid pharmaceutical compositions prepared by dissolving nanocrystalline coatings  
or powders of the antimicrobial metals in pharmaceutically acceptable carriers such as  
water, for application in drop, mist or aerosol forms.

32 In the pharmaceutical compositions, the amount of the nanocrystalline metal powder

1 may range broadly from about 0.001% to about 30% by weight, but will more preferably fall  
2 in the range of from about 0.01 to 5% by weight. Coatings of the nanocrystalline  
3 antimicrobial metals may be very thin, or thicker, depending on the desired duration of  
4 application on the patient. Typical coating thicknesses are in the range of 150 to 3000 nm  
5 thick. As liquid formulations, the amount of dissolved antimicrobial metal will typically  
6 range between about 0.001 to 1% by weight.

7 Besides the active ingredient, pharmaceutical compositions may also include non-  
8 toxic, pharmaceutically and dermatologically acceptable carriers, diluents and excipients,  
9 suitable for topical application, as are well known, see for example Merck Index, Merck &  
10 Co., Rahway, N.J., Bioreversible Carriers in Drug Design, Theory and Application, Roche  
11 (ed.) Pergamon Press, (1987), Gilman et al., (eds) (1990) Goodman and Gilman's: The  
12 Pharmacological Bases of Therapeutics, 8<sup>th</sup> Ed., Pergamon Press; Novel Drug Delivery  
13 Systems, 2<sup>nd</sup> Ed., Norris (ed.) Marcel Dekker Inc., (1989), and Remington's Pharmaceutical  
14 Sciences. For standard dosages of conventional pharmacological agents, see, e.g., Physicians  
15 Desk Reference (1997 Edition); and American Medical Association (1997) Drug Evaluations  
16 (Subscriptions).

17 Dosage forms for the topical administration of compositions of the nanocrystalline  
18 antimicrobial metals include various mixtures and combinations that can be applied topically  
19 and will permit even spreading and absorption into the cutaneous surfaces. Examples include  
20 sprays, mists, aerosols, lotions, creams, solutions, gels, ointments, pastes, emulsions, and  
21 suspensions. The active compound can be mixed under sterile conditions with a  
22 pharmaceutically acceptable carrier, and with any preservatives, buffers, or propellants which  
23 may be required. Topical preparations can be prepared by combining the antimicrobial metal  
24 powder with conventional pharmaceutically acceptable diluents and carriers commonly used  
25 in topical dry, liquid, cream and aerosol formulations. Ointment and creams can, for  
26 example, be formulated with an aqueous or oily base with the addition of suitable thickening  
27 and/or gelling agents. An exemplary base is water. Thickening agents which can be used  
28 according to the nature of the base include aluminum stearate, cetostearyl alcohol, propylene  
29 glycol, polyethylene glycols, hydrogenated lanolin, and the like. Lotions can be formulated  
30 with an aqueous base and will, in general, also include one or more of the following:  
31 stabilizing agents, emulsifying agents, dispersing agents, suspending agents, thickening  
32 agents, coloring agents, perfumes, and the like. Powders can be formed with the aid of any

1       suitable powder base, e.g., talc, lactose starch and the like. Drops can be formulated with an  
2       aqueous base or non-aqueous base, and can also include one or more dispersing agents,  
3       suspending agents, solubilizing agents, and the like.

4           Ointments, pastes, creams and gels also can contain excipients, such as starch,  
5       tragacanth, cellulose derivatives, polyethylene glycols, silicones, bentonites, silicic acid, and  
6       talc, or mixtures thereof. Powders and sprays also can contain excipients such as lactose,  
7       talc, silicic acid, aluminum hydroxide, calcium silicates and polyamide powder, or mixtures  
8       of these substances. Solutions of nanocrystalline antimicrobial metals can be converted into  
9       aerosols or sprays by any of the known means routinely used for making aerosol  
10      pharmaceuticals. In general, such methods comprise pressurizing or providing a means for  
11      pressurizing a container of the solution, usually with an inert carrier gas, and passing the  
12      pressurized gas through a small orifice. Sprays can additionally contain customary  
13      propellants, such as chlorofluorohydrocarbons and volatile unsubstituted hydrocarbons, such  
14      as butane and propane.

15           Multiple inactive ingredients are generally incorporated in topical formulations to  
16      improve cosmetic acceptability, and are optional ingredients in the formulations. Examples  
17      of ingredients are emulsifiers, emollients, thickening agents, solvents, colouring agents, anti-  
18      foaming agents, preservatives, fragrances, and fillers may also be added, as is well known in  
19      the art.

20           The dosage of the active ingredients depends upon many factors that are well known  
21      to those skilled in the art, for example, the particular form of the active ingredient, the  
22      condition being treated, the age, weight, and clinical condition of the recipient patient, and  
23      the experience and judgement of the clinician or practitioner administering the therapy. A  
24      therapeutically effective amount of the nanocrystalline antimicrobial metal is that which  
25      provides either subjective relief of symptoms or an objectively identifiable improvement as  
26      noted by the clinician or other qualified observer. The dosing range varies with the metal  
27      used, its form, the route of administration and the potency of the particular compound.

28      D. Methods of Treating Acne

29           The invention provides methods of treating acne, by administering a therapeutically  
30      effective amount of a nanocrystalline antimicrobial metal powder, or a solution derived from  
31      a nanocrystalline antimicrobial metal, as either a topical formulation, or as a coating on  
32      medical dressing, applied to the locally affected skin area. A therapeutically effective amount

1 may be determined by testing formulations containing the nanocrystalline antimicrobial  
2 metals by *in vitro* or *in vivo* testing. Topical applications may be applied one or more times a  
3 day. Dressings coated with the nanocrystalline antimicrobial metals may be changed daily, or  
4 even less frequently, and should be kept in a moist condition with the addition of saline,  
5 alcohols, or more preferably sterile water, in order to release ions, atoms, molecules or  
6 clusters of the nanocrystalline metal, on a sustained basis.

7 **E. Examples**

8 **Example 1 - Preparation of Nanocrystalline Silver Coatings on Dressings**

9 This example shows the preparation of a bilayer nanocrystalline silver coating on a  
10 dressing material. A high density polyethylene dressing, DELNET™ or CONFORMANT  
11 2™ was coated with a silver base layer and a silver/oxide top layer to generate a coloured  
12 anti-microbial coating having indicator value. The coating layers were formed by magnetron  
13 sputtering under the conditions set out in Table 1.

14 **Table 1**

<u>Sputtering Conditions:</u>	<u>Base Layer</u>	<u>Top Layer</u>
Target	99.99% Ag	99.99% Ag
Target Size	20.3 cm diameter	20.3 cm diameter
Working Gas	96/4 wt% Ar/O <sub>2</sub>	96/4 wt% Ar/O <sub>2</sub>
Working Gas Pressure	5.33 Pa (40 mT)	5.33 Pa (40 mT)
Power	0.3 kW	0.15 kW
Substrate Temperature	20°C	20°C
Base Pressure	3.0 X 10 <sup>-6</sup> Torr	3.0 X 10 <sup>-6</sup> Torr
Anode/Cathode Distance	100 mm	100 mm
Sputtering Time	7.5 - 9 min	1.5 min
Voltage	369 - 373 V	346 V

26 The resulting coating was blue in appearance. A fingertip touch was sufficient to  
27 cause a colour change to yellow. The base layer was about 900 nm thick, while the top layer  
28 was 100 nm thick.

29 To establish that silver species were released from the coated dressings, a zone of  
30 inhibition test was conducted. Mueller Hinton agar was dispensed into Petri dishes. The agar  
31 plates were allowed to surface dry prior to being inoculated with a lawn of *Staphylococcus*  
32 *aureus* ATCC No. 25923. The inoculant was prepared from Bactrol Discs (Difco, M.), which  
33 were reconstituted as per the manufacturer's directions. Immediately after inoculation, the  
34 coated materials to be tested were placed on the surface of the agar. The dishes were  
35 incubated for 24 hr. at 37°C. After this incubation period, the zone of inhibition was

1 calculated (corrected zone of inhibition = zone of inhibition - diameter of the test material in  
2 contact with the agar). The results showed a corrected ZOI of about 10 mm, demonstrating  
3 good release of silver species.

4 The coating was analyzed by nitric acid digestion and atomic absorption analysis to  
5 contain 0.24 +/- 0.04 mg silver per mg high density polyethylene. The coating was a binary  
6 alloy of silver (>97%) and oxygen with negligible contaminants, based on secondary ion  
7 mass spectroscopy. The coating, as viewed by SEM, was highly porous and consisted of  
8 equiaxed nanocrystals organized into coarse columnar structures with an average grain size of  
9 10 nm. Silver release studies in water demonstrated that silver was released continuously  
10 from the coating until an equilibrium concentration of about 66 mg/L was reached  
11 (determined by atomic absorption), a level that is 50 to 100 times higher than is expected  
12 from bulk silver metal (solubility < 1mg/L).

13 By varying the coating conditions for the top layer to lengthen the sputtering time to 2  
14 min, 15 sec., a yellow coating was produced. The top layer had a thickness of about 140 nm  
15 and went through a colour change to purple with a fingertip touch. Similarly, a purple  
16 coating was produced by shortening the sputtering time to 1 min, to achieve a top layer  
17 thickness of about 65 nm. A fingertip touch caused a colour change to yellow.

18 To form a three layer dressing, two layers of this coated dressing material were  
19 placed above and below an absorbent core material formed from needle punched  
20 rayon/polyester (SONTARA<sup>TM</sup> 8411). With the silver coating on both the first and third  
21 layers, the dressing may be used with either the blue coating side or the silver side in the skin  
22 facing position. For indicator value, it might be preferable to have the blue coating visible.  
23 The three layers were laminated together by ultrasonic welding to produce welds between all  
24 three layers spaced at about 2.5 cm intervals across the dressing. This allowed the dressing to  
25 be cut down to about 2.5 cm size portions for smaller dressing needs while still providing at  
26 least one weld in the dressing portion.

27 The coated dressings were sterilized using gamma radiation and a sterilization dose of  
28 25 kGy. The finished dressing was packaged individually in sealed polyester peelable  
29 pouches, and has shown a shelf life greater than 1 year in this form. The coated dressings can  
30 be cut in ready to use sizes, such as 5.1 x 10.2 cm strips, before packaging. Alternatively, the  
31 dressings may be packaged with instructions for the patient or clinician to cut the dressing to  
32 size.

- 1        Additional silver coated dressings were prepared in a full scale roll coater under  
2        conditions to provide coatings having the same properties set out above, as follows:  
3           • the dressing material included a first layer of silver coated DELNET, as set  
4                out above, laminated to STRATEX, AET, 8.0NP<sub>2</sub>-A/QW, which is a layer of  
5                100% rayon on a polyurethane film.  
6           • Silver Foam Dressing - three layers of silver coated high density polyethylene  
7                prepared as above, alternating with two layers of polyurethane foam, L-  
8                00562-6 Medical Foam, available from Rynel Ltd., Bootbay, Maine, USA.

9        Example 2 - Preparation of Nanocrystalline Silver Coating on HDPE Mesh

10       The silver coated mesh was produced, as set forth in Example 1, by sputtering silver  
11       onto Delnet, a HDPE mesh (Applied Extrusion Technologies, Inc., Middletown, DE, USA)  
12       using Westaim Biomedical TMRC unit under the following conditions:

13       Table 2 - Sputtering conditions

14       Target:	99.99% Ag
15       Target Size:	15.24 cm X 152.4 cm
16       Working Gas:	99.375:0.625 wt% Ar/O <sub>2</sub>
17       Working Gas Pressure:	5.33 Pa (40 mTorr)
18       Total Current:	22 A
19       Base Pressure:	5.0 X 10 <sup>-5</sup> Torr
20       Web Speed:	577 mm/min
21       Voltage:	367 V

22       The coating was tested and found to have a weight ratio of reaction product to silver  
23       of between 0.05 and 0.1. The dressing was non-staining to human skin.

24       Example 3 - Preparation of Nanocrystalline Silver Powders

25       Nanocrystalline silver powder was prepared by sputtering silver coatings on silicon  
26       wafers using Westaim Biomedical NGRC unit, and then scraping the coating off. The  
27       sputtering conditions were as follows:

28       Table 3 - Sputtering Conditions

29       Target:	99.99% Ag
30       Target Size:	15.24 cm X 1216.125 cm
31       Working Gas:	75:25 wt% Ar/O <sub>2</sub>

1 Working Gas Pressure: 40 mTorr  
2 Total Current: 40 A  
3 Base Pressure: 5.0 X10<sup>-5</sup> Torr  
4 Sandvik Belt Speed: 340 mm/min  
5 Voltage: 370 V

6 The powder has a particle size ranging from 2  $\mu\text{m}$  to 100  $\mu\text{m}$ , with crystallite size of 8  
7 to 10 nm, and demonstrated a positive rest potential.

8 Example 4 - In vitro Activity of Silver Solution against *Propionibacterium acne*:

9 An *in vitro* test was conducted to determine if silver solutions according to the present  
10 invention effectively control *Propionibacterium acne*. The silver solution was obtained by  
11 static elution of Acticoat™ Burn Wound Dressing (lot #: 00403A-05, Westaim Biomedical  
12 Corp., Fort Saskatchewan, Canada) with nanopure water in a ratio of one square inch of  
13 dressing in five milliliters of water for 24 hours at room temperature. The silver concentration  
14 of the silver solution was determined by an atomic absorption method. The silver elute was  
15 diluted with nanopure water to 20  $\mu\text{g}/\text{ml}$ . The *Propionibacterium acne* (ATCC No. 0919)  
16 was provided by Biofilm Research Group, University of Calgary.

17 The inoculum was prepared by inoculating freshly autoclaved and cooled tubes of  
18 Tryptic soy broth (TSB) with *P. acnes* and incubating them for 2 days at 37°C in an anaerobic  
19 jar. At this time, the optical density of the suspensions was ~ 0.3 at a wavelength of 625 nm.

20 The bacterial suspension (100  $\mu\text{L}$ ) was mixed with 100  $\mu\text{L}$  of the silver solution being  
21 tested. The final concentration of silver in these mixtures was 10  $\mu\text{g}/\text{ml}$ . The mixtures were  
22 incubated in an anaerobic jar at 37°C for two hours. The silver was neutralized by addition of  
23 0.4% STS (0.85%NaCl, 0.4% Sodium thioglycolate, 1% Tween 20) and the solution was  
24 serially 10-fold diluted with phosphate-buffered saline. 20  $\mu\text{L}$  aliquots of the original  
25 solution and subsequent dilutions were plated onto TSA drop plates. The drops were allowed  
26 to dry and the plates were incubated in an anaerobic jar at 37°C for 72 hours at which time the  
27 colonies were counted. The control consisted of 100  $\mu\text{L}$  of bacterial suspension mixed with  
28 100  $\mu\text{L}$  of nanopure water and treated as above.

29 The results showed that the silver solution according to the present invention, at a  
30 final concentration of 10  $\mu\text{g}/\text{ml}$ , gave 4.3 logarithm reduction in viable *Propionibacterium*  
31 *acne* counts in two hours.

1      Example 5 - Treatment of Acne

2      A sixteen year old female was diagnosed with acne vulgaris. She had numerous red  
3      papules and pustules on her forehead. Various skin cleansing regimes and antibiotic  
4      (erythromycin and clindomycin) treatments had been tried and had failed to control the acne.  
5      Prior to bed time, the papules and pustules on one side of her forehead were moistened and  
6      covered with a nanocrystalline silver coated high density polyethylene mesh, prepared as in  
7      Example 1 (single layer, blue coating). The mesh was then occluded with a thin film dressing  
8      which remained in place for 10 hours. Upon removal the papules and pustules were no  
9      longer red and were only slightly raised. Some brown staining of the skin was observed.

10     Example 6 - Treatment of Acne

11     A sixteen year old male was diagnosed with acne vulgaris. He had numerous raised,  
12     red papules and pustules on his forehead. Various skin cleansing regimes and antibiotic  
13     treatments had been tried and had failed to control the acne. The patient was placed on  
14     isotretinoin treatment which controlled his acne well. He did develop a single large pustule  
15     on his forehead which was embarrassing for him. Prior to bed time, the pustule was  
16     moistened and covered with a nanocrystalline silver coated high density polyethylene mesh  
17     prepared as in Example 2.. The mesh was then occluded with a thin film dressing which  
18     remained in place for 10 hours. Upon removal the pustule was no longer red and was only  
19     slightly raised. A second treatment resulted in the disappearance of the pustule.

20     Example 7 - Treatment of Acne

21     A sixteen year old female was diagnosed with acne vulgaris. She had numerous red  
22     papules and pustules on her forehead. Various skin cleansing regimes and antibiotic  
23     (erythromycin and clindomycin) treatments had been tried and had failed to control the acne.  
24     Prior to bed time, the papules and pustules on one side of her forehead were moistened and  
25     covered with a nanocrystalline silver coated high density polyethylene mesh, prepared as in  
26     Example 2. The mesh was then occluded with a thin film dressing which remained in place  
27     for 10 hours. Upon removal the papules and pustules were no longer red and were only  
28     slightly raised. A second treatment resulted in the disappearance of the papules and virtual  
29     elimination of the pustules.

30     The silver coated mesh, when prepared as set forth in Example 2, did not result in any  
31     staining of the skin.

32     All publications mentioned in this specification are indicative of the level of skill of

1       those skilled in the art to which this invention pertains. All publications are herein  
2       incorporated by reference to the same extent as if each individual publication was specifically  
3       and individually indicated to be incorporated by reference.

4           The terms and expressions in this specification are, unless otherwise specifically  
5       defined herein, used as terms of description and not of limitation. There is no intention, in  
6       using such terms and expressions, of excluding equivalents of the features illustrated and  
7       described, it being recognized that the scope of the invention is defined and limited only by  
8       the claims which follow.

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